

Magnetic Dynamics of Iron-Oxide Nanoparticles in Frozen Ferrofluids and Ferronematics

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Abstract. We present a detailed study of the magnetic behavior of iron oxide (γ -Fe₂O₃ and Fe₃O₄) nanoparticles constituents of ferrofluids (FF's) with average particle sizes $\langle d \rangle = 2.5$ and 10 nm. The particles were dispersed in the frozen liquid carrier (pure FF) and in a frozen lyotropic liquid crystalline matrix in the nematic phase or ferronematic (FN) (ferrolyomesophase). Both FF and FN phases displayed superparamagnetic (SPM) behavior at room temperature, with blocking temperatures $T_B \sim 10$ and 100 K for $\langle d \rangle = 2.5$ and 10 nm, respectively. Dynamic ac susceptibility measurements showed a thermally activated Néel-Brown dependence of the blocking temperature with applied frequency. Our results show that dipolar interactions are small, but non-negligible, as compared to the single-particle energy barriers from magnetic anisotropy. From the fit of ac susceptibility we calculated the effective magnetic anisotropy constant K_{eff} for 2.5 nm maghemite particles. Although interparticle interactions present in highly diluted samples do not appreciably modify the dynamic magnetic behavior of isolated particles, the calculated magnetic anisotropy were about one order of magnitude larger than the bulk materials, suggesting the existence of large surface anisotropy. Using the thermally activated model to fit the dynamic data yielded effective energy barriers $E_a = 3.5 \times 10^{-21}$ J. From these data, we obtained $K_{\text{eff}} = 422$ kJ/m³ for the single-particle effective magnetic anisotropy.

Introduction

Nanostructures, a size regime between molecular and submicron-size structures, possess high surface-to-volume ratios, yielding novel properties. Many biological and biomedical applications use such nanoscopic systems in a magnetic fluid or *ferrofluid*, a colloidal suspension of magnetic oxide nanograins dispersed in a liquid carrier. Depending on the characteristic of the liquid carrier (polar or apolar character and pH), magnetic grains are coated with an appropriate agent for stability purposes. The synthesis and applications of magnetic fluids for biomedical applications have passed the speculative stage of development, to enter into the well-based standard laboratory procedures. Nowadays, these particles are being used in magnetic resonance contrast enhancement, cell and DNA separation, drug delivery, and gene cloning. Often, the desired properties include large magnetic moments per formula unit, making iron oxide particles such as Fe₃O₄ and γ -Fe₂O₃ appealing for these applications. For biocompatible purposes, particles are usually coated with noble metals (Gd, Pt, Ag), or organic substances such as dextran. [1] In spite of the above advancements on bio-magnetic applications, there is still a lack of precise knowledge on the mechanisms linking particle shape, size distribution and surface structure, to the resulting magnetic properties of the carrier magnetic nanoparticles.

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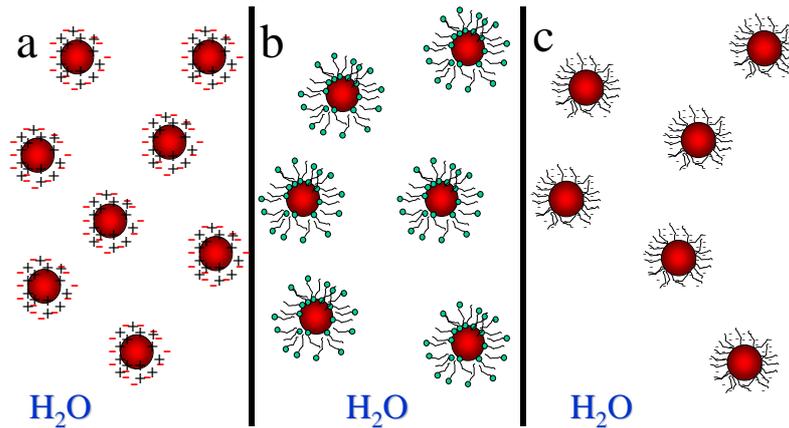


Fig. 1. Illustrative picture of different coatings for stabilize the magnetic grains in a particular liquid carrier: Ionic (a), surfacted (b), and citrated (c) types.

The lyotropic liquid crystals as those used in this work are organized molecular systems formed by a mixture of amphiphilic molecules (molecules with both hydrophobic and hydrophilic parts) and a solvent, most often water, and their structure depends on temperature and relative composition of the constituents [2]. Amphiphilic aggregates in the nematic mesophase, named *micelles* (typical dimensions 10 nm), have a bilayer structure, resembling cell membranes.

Regarding the magnetic grains, both maghemite and magnetite have a cubic spinel structure with two interacting magnetic sublattices. The strong superexchange coupling between A and B magnetic sublattices yields antiparallel alignment, and thus ferrimagnetic order below T_C . The first-order magnetocrystalline anisotropy constants are (at room temperature) $K_1 = 13.5 \text{ kJ/m}^3$ and 4.7 kJ/m^3 for Fe_3O_4 and $\gamma\text{-Fe}_2\text{O}_3$, respectively. Interparticle interactions can modify the particle energy barriers, yielding a shift in the observed blocking temperature [3] due to collective freezing of the magnetic moments. Concurrently with dipolar interactions, other sources of magnetic anisotropy (e.g. surface effects) can have a major impact on the blocking process. In this scenario the freezing process, analyzed as a superspin-glass transition, is subject of intense debate. In this work, we present a magnetic characterization of magnetite (Fe_3O_4) and maghemite ($\gamma\text{-Fe}_2\text{O}_3$) nanoparticles with mean size $\langle d \rangle$ of 10 nm and 2.5 nm, respectively, with different concentrations to analyze the effects of dipolar interactions on the magnetic dynamic of the constituent particles.

Experimental Procedure

We have studied the magnetic interaction between the nanograins in frozen samples of a pure FF and a FF-doped lyotropic liquid crystals in the nematic phase. The magnetic particles of the starting FF were coated with a surfactant (figure 1) to prevent agglomeration, and suspended in a nonmagnetic solvent as illustrated in Fig. 1. Ferrofluids are magnetic colloids in which magnetic oxide nanograins are dispersed in a liquid carrier. The two main types used in this work were surfacted (SFF) and citrated (CFF) ferrofluids (see fig. 1). Magnetite (Fe_3O_4) particles having $\langle d \rangle = 10 \text{ nm}$ were studied from a commercial SFF from Ferrotec[®] Company, whereas the $\langle d \rangle = 2.5 \text{ nm}$ maghemite ($\gamma\text{-Fe}_2\text{O}_3$) particles were from a non-commercial CFF. The ferronematic (FN) phases were obtained by mixing the corresponding FF with lyotropic crystals with different concentrations. The relative volume concentration of the samples, $R_V = 100 \times \frac{\text{vol. magnetic phase}}{\text{total volume}}$ is shown in Table 1.

Static and dynamic magnetic measurements as a function of frequency and temperature were performed in a commercial SQUID magnetometer (Quantum Design). Zero-field-cooled (ZFC) and field-cooled (FC) curves were taken between 5 K and 250 K, for different values of cooling field H_{FC} ($10 \text{ Oe} < H < 70 \text{ kOe}$). AC susceptibility of the samples was obtained for frequencies from 10^{-2} to

10^3 Hz, as a function of temperature. Magnetization data were obtained by first cooling the sample from room temperature in zero applied field (ZFC process) to the basal temperature. Then a field was applied and the variation of magnetization was measured with increasing temperature up to $T = 250$ K. After the last point was measured, the sample was cooled again to the basal temperature keeping the same field (FC process); then the M vs. T data was measured for increasing temperatures.

Label	Sample	$\langle d \rangle$ (nm)	$R_V\%$ (vol.)	T_B (K)	$H_C(5K)$ (Oe)	$H_C(30K)$ (Oe)	M_R/M_S (5K)	M_R/M_S (30K)
SFF0	Commercial Ferrofluid (SFF)	10	1.8	218	80(5)	70(5)	0.12(2)	0.12(2)
SFFC	Concentrated SFF Ferronematic	10	$\sim 10^{-2}$	228	125(10)	103(5)	--	--
SFFD	Diluted SFF Ferronematic	10	$\sim 2 \times 10^{-4}$	169	100(10)	98(8)	--	--
CFF0	Non-commercial Ferrofluid (CFF)	2.5	1.1	9.1	55 (5)	12(5)	0.06(1)	0.00(1)
CFFC	Concentrated CFF Ferronematic	2.5	0.257	13.9	115(12)	5(5)	--	--
CFFD	Diluted CFF Ferronematic	2.5	$\frac{0.0007}{6}$	11.1	70(10)	5(5)	--	--
CFFY	Dried CFF Ferrofluid	2.5	Unkno wn	11.8	65(5)	13(5)	0.07(1)	0.01(1)

Table 1. Description, structural and magnetic parameters of the samples studied in this work: mean particle diameter $\langle d \rangle$, volumetric concentration R_V (%), Blocking temperature T_B (at $H = 100$ Oe), coercive fields H_C and remanence to saturation ratio M_R/M_S .

Results and Discussion

The structural and magnetic parameters of the studied samples are condensed in Table 1. The different average particle diameter for the two SFF and CFF samples are reflected in the blocking temperatures T_B (Fig. 2). For the SFF samples, the ZFC magnetization curves showed a broad maximum at ~ 200 K, which extends above the freezing temperature of the matrix (~ 260 K). Therefore, at our maximum measuring temperatures of ca. 250 K, some of the particles might be still in the blocked state. For the CFF samples, the T_B values vary between 9(1) and 14(1) K, depending on the particle concentration. For these samples, there will be no significant blocked particles at ~ 250 K. Indeed, the coercive fields at $T = 30$ K for SFF and CFF samples show this difference (Table 1 and Fig. 2): whereas samples CFF have essentially zero coercive field H_C (5-10 Oe), for SFF samples a significant H_C (70-100 Oe) is observed due to a blocked fraction of the particles. Interestingly, the blocking temperatures displayed in Table 1 show a measurable increase from the FF phase to the FN concentrated and diluted. We have checked the effect of interparticle interaction by evaporating a large fraction of the solvent in the CFF sample, obtaining a 'dried' sample (CFFY) where particles are closer to each other. It can be observed that for this sample, T_B shifts upwards by $\sim 30\%$ of the CFF0 sample, indicating that the effect of dipolar interactions goes in this direction. Since the average interparticle distance in FN phases is much larger (high dilution) we attribute the observed increase in T_B to particle agglomeration formed in these samples. As the concentration used were well above the standards for colloidal stability, this effect is likely to occur.

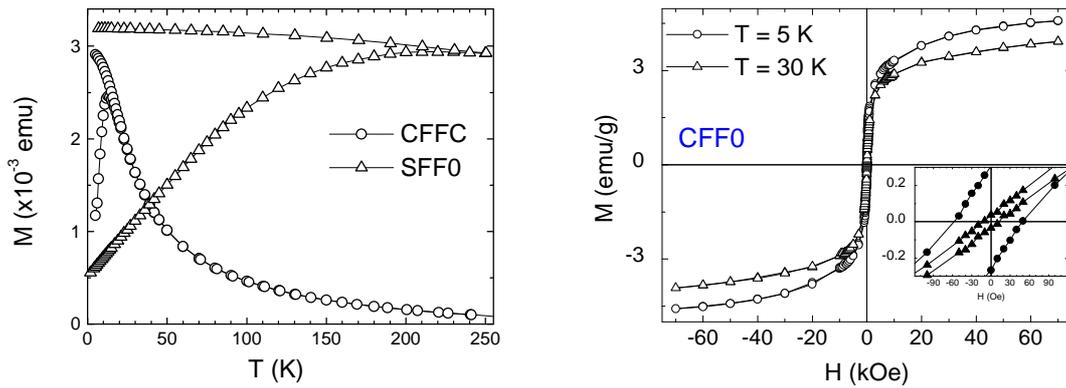


Fig. 2. Left panel: AFC/FC curves for surfacted ($\langle d \rangle = 10$ nm) and citrated ($\langle d \rangle = 2.5$ nm) ferrofluids, showing their respective blocking temperatures. Right panel: Typical magnetization curves as a function of applied field.

For each set of samples (CFF and SFF), a slight increase in H_C can be observed from the pure ferrofluid to the ferronematic phases. Table 1 also displays the remanence-to-saturation ratio M_R/M_S . For the FN samples, this ratio has not clear meaning due to the diamagnetic contribution of the matrix, which dominates at high fields. However, for the measured ratios the observed value is much smaller than the expected. At $T = 5$ K, the M_R/M_S values for the samples studied (see Table 1) are much smaller than the expected $R = 0.5$ value for noninteracting, randomly oriented particles with uniaxial symmetry. Deviations from the non-interacting values are expected, and moreover these low values have been related to the existence of inter-particle interactions of antiferromagnetic nature. [4]

We have investigated the magnetic dynamics of the $\langle d \rangle = 2.5$ nm ferrofluid samples, in order to detect the effect of interparticle interactions on the single-particle behavior. Fig. 3 displays the temperature dependence of $\chi'(T)$ and $\chi''(T)$ of the smallest particles (samples CFF0 and CFFY) for different frequencies f ranging from 20 mHz to 1 kHz. The data for both components $\chi'(T)$ and $\chi''(T)$ exhibit the expected behavior of SPM systems, i.e., the occurrence of a maximum at a temperature T_m ($\sim T_B$) for both $\chi'(T)$ and $\chi''(T)$ components, which shifts towards higher values with increasing frequency. [3] From the empirical relationship $\Phi = \Delta T_B / T_B [\Delta \log_{10}(f)]$, where ΔT_B is the difference between T_B measured in the $\Delta \log_{10}(f)$ frequency interval, we obtained $\Phi = 0.089(2)$ and $0.067(2)$ for CFF0 and CFFY, respectively, indicating the presence of interparticle interactions the usually yield Φ values lower than the 0.1-0.13 theoretically expected. [5,6] Accordingly, the effect of dipolar interactions in the more concentrated sample (CFFY) is reflected in the lower Φ value. It is well known that the dynamic response of an ensemble of fine particles is determined by the measuring time τ_m (or frequency) of each experimental technique. As the reversion of the magnetic moment in a single-domain particle over the anisotropy energy barrier E_a is assisted by thermal phonons, the relaxation time τ exhibits an exponential dependence on temperature characterized by

$$t = t_0 \exp\left(\frac{E_a}{k_B T}\right) \quad (1)$$

where τ_0 is in the $10^{-9} - 10^{-11}$ s range for SPM systems. In the absence of an external magnetic field, the energy barrier can be assumed to be $E_a = K_{\text{eff}} V$, where the proportionality constant K_{eff} is an effective magnetic anisotropy.

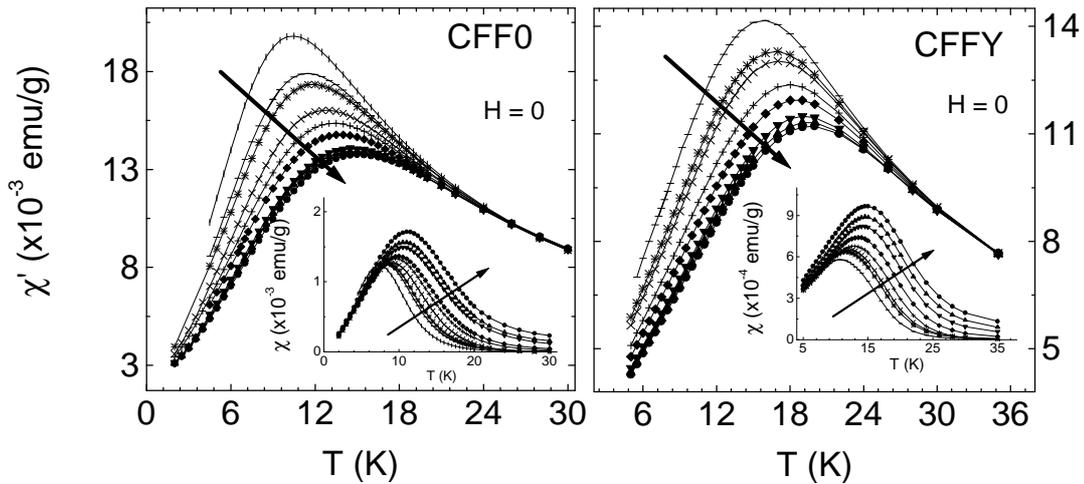


Fig. 3. Temperature dependence of the in-phase (real) component $\chi'(T)$ of the magnetic susceptibility for CFF samples, at different excitation frequencies, taken in zero external magnetic field. The arrows indicate increasing frequencies. Inset: Out of phase (imaginary) component $\chi''(T)$.

The linear dependence of $\ln(f)$ versus $1/T_B$ observed in Fig. 4 indicates that the Néel-Brown model correctly suits the behavior of CFF samples. From the fitting of the experimental $T(f, H=0)$ data using Eq. 1, we obtained the values of $E_a/k_B = 250(7)$ K for CFF0 sample and $E_a/k_B = 412(11)$ K for the more concentrated sample CFFY. This increment in E_a agrees with the expected contribution from dipolar interactions to the single-particle effective energy barrier. Using the average particle diameter $\langle d \rangle = 2.5$ nm, we obtained $K_{\text{eff}} = 422$ kJ/m³ and 695 kJ/m³ for the effective anisotropies for samples CFF0 and CFFY, respectively.

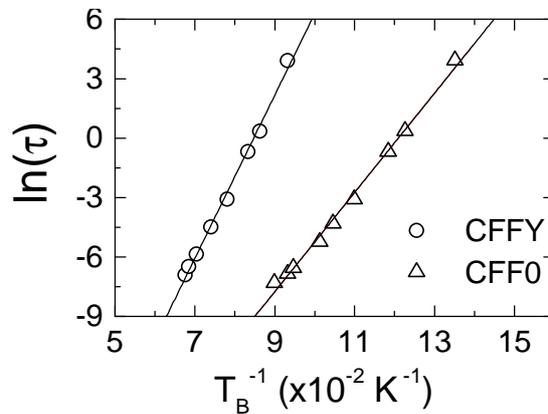


Fig. 4. Arrhenius plot of the relaxation time τ vs. T_B^{-1} obtained from the imaginary component $\chi''(T)$ of figure 3. Solid line is the best fit using Eq. (1).

The larger E_a value found in concentrated sample should be attributed to the increase of dipolar interactions that add to the single particle energy barrier. However, the K_{eff} value calculated for the most diluted sample is approximately two orders of magnitude larger than the magnetocrystalline anisotropy constant of bulk maghemite $K_1^{\text{bulk}} = 4.7$ kJ/m³, indicating the existence of additional anisotropy sources. In previous works on more concentrated nanoparticle systems, large values of particle anisotropy are usually related to surface effects and/or the effect of dipolar interactions [3].

The contribution from particle-particle interactions to the experimental data is difficult to disentangle from the single-particle magnetic anisotropy. In addition to the intrinsic sources of particle anisotropy (e.g. shape, stress and magnetocrystalline anisotropies), interparticle interactions

(mainly the long-range dipolar interaction) can also modify the energy barrier values. In a previous work on a ferrofluid consisting in Fe_3O_4 particles of $d = 5\text{nm}$, [7] it was found that dipolar interactions were already noticeable for concentrations R_V of $\sim 2\%$ vol. of magnetic particles. However, the high dilution of our samples (except for the dried sample, see Table 1) guarantees that the value of K_{eff} obtained should be essentially the single-particle value. Surface effects have been invoked as a large source of magnetic anisotropy in Cobalt [8] and maghemite [9] nanoparticles. Our data clearly indicates that both the similarity between K_{eff} for liquid and dried samples and the large K_{eff} value are a consequence of an intrinsic property of these maghemite particles, suggesting that surface anisotropy is dominant.

Summary

We have studied the effect of interparticle interactions on the magnetic properties of single-domain Fe_3O_4 and $\gamma\text{-Fe}_2\text{O}_3$ nanoparticles with identical size distribution but different concentration. We found that stronger interparticle interactions yield an increase of T_B , as observed from agglomeration of particles in FN phases. For diluted ferrofluid samples with maghemite particles having $\langle d \rangle = 2.5\text{ nm}$, the calculated anisotropy constant is much larger than the bulk value, indicating a large contribution to the magnetic anisotropy, which is likely to be located at the particle surface.

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